

Cobalt-Free Cathodes for Next Generation Lithium-Ion Batteries

Principal Investigator: Neil J. Kidner

Project ID: bat417



OVERVIEW

Budget

Total Project Funding: \$3.08 M

- DOE share: \$2.46 M
- Cost share: \$620 k

Funding for FY2020: \$1.1 M
Funding for FY2021: \$1.0 M
Funding for FY2022: \$1.0 M

Timeline

Project Start Date: Jan. 2019
Project End Date: Mar. 2022
Percent Complete: 100 %

Partners

Ohio State University: Dr. Jung-Hyun Kim
Battery testing
Cell chemistry development
Navitas Systems Dr. James Dong
Large-scale electrode fabrication
2-Ahr battery manufacture and testing

Barrier and Technical Targets

Cycle Life: 1000 cycles C/3 deep discharge with < 20 % energy fade
Cost: < \$100/kWh

RELEVANCE

Impact

The increased demand for EVs is driving demand for battery materials.
Renewed interest in reduced/cobalt free Li-ion battery cathode formulations

- Opportunity to reestablish U.S. dominance in batteries

Objective

Develop high performance and cobalt-free Li-ion battery based on high voltage lithium manganese nickel titanium oxide (LNMTMO) cathode and complementary cell chemistry (electrolyte/cathode formulation)
Identify low-cost, scalable process for producing cathode powder
Identify strategy for Nexceris to grow domestic manufacturing and create jobs to support new clean energy and e-mobility opportunities

APPROACH

TiOx-enriched inorganic SEI

- Self healing
- In-situ formation
- Passivation of active material

Core-shell microstructure

- TiOx enriched surface shell
- Low Ti substitution in core

LIPAA polymeric SEI

- In-situ formation
- Passivation of active material and carbon
- Extra Li+ donation
- Proton scavenge

Aluminum

Carbon + Binder

Electrolyte

PF5

HF

LNMTMO

Develop cobalt-free cell based on high-voltage $\text{LiNi}_{0.5}\text{Mn}_{1.2}\text{Ti}_{0.3}\text{O}_4$ (LNMTMO) cathode

- Improve cycle and calendar life by forming a solid-electrolyte interface that effectively passivates the cathode surface
- Create novel LNMTMO core/shell microstructures where Ti is preferentially located at surface and partner with optimized binder/electrolyte chemistries

ACKNOWLEDGEMENTS

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This presentation does not contain any proprietary, confidential, or otherwise restricted information

ACCOMPLISHMENT: Core-Shell Cathode Optimization

Building on learnings from Year 2 Intermediate 2-Ah cells the HAWCS synthesis process was refined to optimize the LNMO/LNMTMO core-shell microstructure

Optimized precipitating agent to transition metal ratio during initial precipitation step
Improved LNMO compositional uniformity resulting in higher capacity

Precipitating agent to TM: 4:1 (Year 2 STD)

Precipitating agent to TM: 6:1

Precipitating agent to TM: 8:1

Refined HAWCS synthesis process to target larger diameter core particle
Large core particle reduces surface area and minimizes degradation

Year 1 Solid-state synthesis

Year 2 Co-precipitation

Year 3 HAWCS (core)

Year 3 HAWCS (core-shell)

XPS analysis of Year 3 powders
Synthesized by HAWCS
Ar-sputtering was performed from 0 – 15 min.
Deeper Ti penetrations with increasing Ti in the shell compositions:

- Bare LNMO: Ti-free
- LNMTMO-1: $\text{LiNi}_{0.5}\text{Mn}_{1.2}\text{Ti}_{0.15}\text{O}_4$
- LNMTMO-2: $\text{LiNi}_{0.5}\text{Mn}_{1.2}\text{Ti}_{0.3}\text{O}_4$
- LNMTMO-3: $\text{LiNi}_{0.5}\text{Mn}_{1.2}\text{Ti}_{0.5}\text{O}_4$

HAWCS development provided a systematic improvement in performance for both LNMO core and LNMO/LNMTMO core-shell cathodes during the project

Specific Capacity (mAh/g)

Year-1 LNMO
Year-1 LNMTMO
Year-2 LNMO
Year-2 LNMTMO
Year-3 LNMO
Year-3 LNMTMO-1
Year-3 LNMTMO-2
Year-3 LNMTMO-3

Single-layer pouch cell tests
Cathode: LNMO/LNMTMO LIPAA binder, Anode: MCMC, Electrolyte: 1 M LIPF6 in 1:1 wt. EC/EMC
Cycle 1,2,3: C/10 rate, rest C/3 @ 25°C

Raman Intensity

Fresh LNMO
Cycled LNMO
Fresh LNMTMO-3
Cycled LNMTMO-3

Raman Shift (cm⁻¹)

Formation of Ti-rich $\text{LiNi}_{0.5}\text{Ti}_{0.3}\text{O}_4$ was found after cycling LNMTMO-3, indicating sacrificial Mn dissolution and formation of the more-stable Ti-enriched CEI layer

Cycling Number

R_{ct} (Ω)
 R_{s+r} (Ω)

Reduced interfacial impedance (R_{ct}) and charge-transfer (R_{s+r}) resistance with Ti-enriched core-shell cathodes in single-layer pouch cells.

ACCOMPLISHMENT: 2 Ah PCC Fabrication and Testing

Throughout Year 3 focused on reducing risk through iterative process development to build confidence in scaled powder production larger cell assembly

Resolved discrepancy in SLP performance – segmented to LiPAA binder shelf life
Old LiPAA binder makes electrode more susceptible to surface cracks

Discharge Capacity (mAh/g)

New LIPAA binder
Old LIPAA binder

SLP Cells Cathode: LNMO/LNMTMO LIPAA binder
Anode: MCMC graphite
Electrolyte 1 M LIPF6 in 1:1 wt. EC/EMC, 4.9-3.5 V, C/3

Cycling Number

Effect of electrode architecture (density, thickness) studied with SLP cells and lower density (higher porosity) electrodes down-selected

Specific Capacity (mAh/g)

Cathode: LNMO, LIPAA binder Anode: MCMC
Electrolyte 1 M LIPF6 in 1:1 wt. EC/EMC (no additive)
Cycle 1,2,3: C/10 rate, rest C/3

Cathode/Anode Porosity

Cycle Number

Discharge Capacity (mAh/g)

Cells 5-8
Cathode density: 1.47 g/cc
Cells 1-4
Cathode density: 1.84 g/cc

1-Ah look-ahead pouch cells
Cathode: LNMO/LNMTMO, LIPAA binder
Anode: MCMC graphite, PVDF binder, 1.21 g/cc density
Electrolyte 1 M LIPF6 in 1:1 wt. EC/EMC (1 wt.% LiBOB additive)
Cycle 3.5-49 V, C/3

Cycle Number

Multiple large LNMO/LNMTMO core-shell powder batches produced with excellent batch-to-batch variation

Specific Capacity (mAh/g)

Batch 1
Batch 2
Batch 3
Batch 4
Blended
final batch

Half-Cell Large batch cathode validation
Cathode: Down-select LNMO Core
Electrolyte: 1 M LIPF6 in 1:1 wt. EC/EMC
Cycle 1,2,3: C/10 rate, rest C/3

Cycle Number

Specific Capacity (mAh/g)

LNMO/LNMTMO Core-Shell
LNMO Core

Half-Cell Large batch cathode validation
Cathode: Down-select LNMO Core and LNMO/LNMTMO core-shell
Electrolyte: 1 M LIPF6 in 1:1 wt. EC/EMC
Cycle 1,2,3: C/10 rate, rest C/3

Cycle Number

Thirty-four 2-Ah Project Completion Cells (PCCs) successfully fabricated; initial C/3 rated capacity characterized and cells delivered to INL, Navitas and OSU

Amp Hours @ C/3 discharge

Cycle 1
Cycle 2
Cycle 3

Cell #

15 best performing 2-Ah PCCs

Energy Wh @ C/3 discharge

Cell #

MILESTONE REVIEW

Milestone	Metric	Status
Down-select PCC LNMO/LNMTMO core shell cathode powder	1. Multiple process refinements completed to optimize the LNMO/LNMTMO core/shell powder Down-selection based on 2-Ah cell testing	Completed <ul style="list-style-type: none">Three LNMO/LNMTMO powder iterations completed and PCC powder down selectedCathode powder shipped to Navitas for PCC fabrication
PCC cell chemistry down-selected	1. Define complementary cell chemistry	Completed <ul style="list-style-type: none">Utilized SLP cell/1-Ah cell formats to identify address cell issues, down-selected cell chemistry/electrode architecture
PCC fabricated	1. Fabricate 30 PCCs to support delivery of 15 to DOE and internal testing.	Completed <ul style="list-style-type: none">LNMO/LNMTMO cathode electrode fabrication completedThirty-four 2-Ah PCCs successfully fabricated
PCC tested	1. Testing of PCCs at INL and Navitas/OSU	Completed <ul style="list-style-type: none">Test plan for PCCs finalized with INLPCCs completed C/3 rated capacity/energy cyclesSub-set of cells cycling at Navitas & OSU

COLLABORATIONS/COORDINATION OTHER INSTITUTIONS

Nexceris is fortunate to have excellent project partners that have supported cell chemistry development, testing and large cell (2-Ah) manufacture and testing.
Nexceris has continued to work to engage with stakeholders throughout the Li-ion EV battery value-chain to identify commercial opportunities for developed technology

Project Team Member	Relationship
	Coin-cell and SLP cell screening of cell chemistries Cell chemistry (additives/binder) development Analytical characterization of cathode materials and electrodes
	Electrode scale-up, formation-cycling optimization Large format 2-Ahr battery fabrication and testing

PROPOSED FUTURE RESEARCH

Future work should focus on addressing the challenges identified during this project and meeting the commercial performance and cost targets for the material
Gas generation, leading to cell-swelling was identified as a major degradation mechanism for large-format cells. The existing formation cycling procedures were insufficient to prevent gas generation.

- Optimization of formation cycling to minimize gas generation
- Detection of electrolyte leakage from swollen/repaired cells

Further optimization of the complementary electrolyte formulation (additives) for high voltage cathode chemistries is also recommended.

SUMMARY

1. In budget period 3 the project team refined the novel Hybrid Alternative Wet-Chemical Synthesis (HAWCS) process for reproducibly producing high-quality LNMO high-voltage cathode powder

2. Three LNMO/LNMTMO core-shell powder iterations were completed, and large-batches of powder delivered to Navitas to address scale-up challenges

3. SLP testing identified LiPAA binder shelf-life and electrode architecture (porosity) as critical allowing these parameters to be optimized

4. LNMO/LNMTMO cathode powder and cell chemistry down-selected and thirty-four 2-Ah cells fabricated